

Biological Evaluation of Some Novel Chalcones and Its Derivatives

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Abstract: CHN analysis Chalcones, (E)-N-(4-(4,6-dichloro-1,3,5-triazin-2-ylamino) phenyl)-3-(4-methoxy-phenyl) acrylamide (4a-c) have been prepared according to Claisen-Schmidt condensation. Further these chalcones (4a-c) on reaction with malononitrile affords cyano-pyridines (5a-c) respectively. The constitutions of newly synthesized compounds have been characterized on the basis of their IR and ¹H NMR, ¹³C NMR spectral data. These synthesized compounds have been screened for their antibacterial and larvicidal activity.

Index Terms: Chalcones, cyanopyridines, Larvicidal activity, antibacterial activity.

I. INTRODUCTION

Heterocyclic compounds are cyclic organic substances which contain at least one atom other than carbon in the ring system. The chemistry of heterocyclic compounds is one of the most complex branches of organic chemistry. A practical method for the synthesis of such compounds is of great interest in synthetic organic chemistry. The chemistry of chalcones have generated intensive scientific studies throughout the world, especially interesting are their biological and industrial applications. Chalcone is a generic term given to compounds bearing the 1,3-diphenyl-2-propen-1-one framework and it belongs to the flavonoid family. Chemically they are open-chain flavonoids in which the two aromatic rings are joined by a three carbon α , β -unsaturated carbonyl system.

Chalcones are readily synthesized in the laboratory by various synthetic methods. Structural modification of chalcone template can be readily achieved. Chalcones are unsaturated ketones containing the reactive keto and ethylenic group $-\text{CO}-\text{CH}=\text{CH}-$ and are colored compounds because of the presence of the chromophore and auxochromes^[1-3]. They are known as benzalacetophenones or benzylideneacetophenones. Kostanecki and Tambor gave the name "Chalcone"^[4-5]. These are found to be effective as anti-inflammatory^[6,7], anticancer^[8-10], antifungal^[11-13], cardiovascular^[14], and antimalarial^[15] agents. The

well-known stepwise reaction between cyanuric chloride and aminoacetanilide is very well defined, and high yields of aminodichlorotriazines were obtained. Cyanuric chloride is definitely an excellent starting compound for the straight forward preparation of highly structured multitopic molecules. The first substitution is exothermic. Therefore, the temperature of the reaction mixture has to be maintained to 0 °C. The substitution of the second step at room temperature, finally the third step is functionalized under reflux of the solvent. These observations led us to synthesize some new s-triazinyl based chalcones and its corresponding cyanopyridine derivatives.

II. EXPERIMENTAL METHODOLOGY

Melting points were determined by Deep Vision instrument. The purity of the compounds was checked by TLC using silica gel coated plates and spots were visualized by exposing the dry plates in iodine vapours. IR spectra were recorded in the solid state, as KBR dispersion by use of the FT-IR-Spectrometer. The ¹H NMR and ¹³C NMR spectra of the compounds were carried out in Bruker AMX 400 MHz. NMR instrument using CDCl₃ or DMSO as a solvent and TMS as internal reference (chemical shift in δ ppm). The mass spectra of the compounds were carried out in ESI Mass.

A. Synthesis of N-(4-(4,6-dichloro-1,3,5-triazin-2-ylamino)phenyl) acetamide (3):

4-amino acetanilide (0.01 mole) was added slowly to cyanuric chloride (0.01 mole) in acetone (30ml) with constant stirring over a period of 4 hr at 0 to 5° C. Then, sodium carbonate (0.05 mole) dissolved in water (10 ml) was added drop wise to neutralize HCl evolved during the reaction. Finally, the contents were poured into crushed ice. The solid was separated out by filtration and washed with water. The product is dried, recrystallized from alcohol to give the product (3).

B. Synthesis of (E)-N-(4-(4,6-dichloro-1,3,5 -triazin-2-ylamino) phenyl)-3-(4-methoxyphenyl) acrylamide (4a):

Acetamide compound (3) (0.01 mole) was dissolved in Ethanol (30 ml) Then 10% NaOH solution and 4-Methoxybenzaldehyde (0.01mole) was added to the reaction mixture with constant stirring over a period of 6 hrs .The reaction mixture was poured into crushed ice. The solid was separated out by filtration and washed with water. The product (4a) is dried , recrystallized from ethanol. IR (KBR) cm^{-1} : -N,s-triazine (829.90), CN-H str (3419.04) , C—Cl (770.81).

C. Synthesis of (E)-N-(4-(4,6-dichloro-1,3,5 -triazin-2-ylamino) phenyl)-3-(4-fluorophenyl) acrylamide (4b):

Acetamide compound (3) (0.01 mole) was dissolved in Ethanol (30 ml) Then 10% NaOH solution and 4-Fluorobenzaldehyde (0.01mole) was added to the reaction mixture with constant stirring over a period of 6 hrs. The reaction mixture was poured into crushed ice. The solid was separated out by filtration and washed with water. The product (4b) is dried , recrystallized from ethanol .IR (KBR) cm^{-1} : C-N,s-triazine (809.95), N-H str (2926.45) , C—Cl (764.63).

D. Synthesis of (E)-N-(3-(4,6-dichloro-1,3,5 -triazin-2-ylamino) phenyl)-3-(benzo [d] [1,3] dioxol-5yl) acrylamide (4c):

Acetamide compound (3) (0.01 mole) was dissolved in Ethanol (30 ml) Then 10% NaOH solution and piperonal (0.01mole) was added to the reaction mixture with constant stirring over a period of 6 hrs .The reaction mixture was poured into crushed ice. The solid was separated out by filtration and washed with water. The product is dried, recrystallized from ethanol. IR (KBR) cm^{-1} : C-N,s-triazine (809.95), N-H str (2922.59) , C—Cl (657.60).

E. Synthesis of 5-(4-(4,6-dichloro-1,3,5 triazin-2ylamino)phenylamino)-2-amino-4-(4-methoxyphenyl) pyridine-3-carbonitrile (5a):

A mixture of a compound (4a) (0.01 mole) dissolved in 40 ml ethanol and added malononitrile (0.01 mole), ammonium acetate (0.08 mole) was refluxed for 8 hrs. Then the mixture was cooled and poured into crushed ice. The product (5a) separated out was filtered washed and recrystallized from alcohol. IR(KBR) cm^{-1} : C-Cl (834.06), Ar C-Cl (1119.48) ,Ar-N str (1383.68), primary N-H (1509.07), C=C (1570.74), C=N (1613.16), N-H str (2853.17). ^1H NMR(CDCl_3) δ ppm : 3.734 (O-CH₃), 4. 311 to 4.349 (S,1H,s-triazine, Ar-C-NH) , 6.986-7.437 (d, 4H,Ar-CH),7.688 (Ar-H), 9.896 (2-Py-Ar-1H). ^{13}C NMR(CDCl_3) δ ppm: Aliphatic-CH₃ (55.51), Ar-CH (119.52 to 121.12), 2-Py (134.85), 1-imine (166.10), S- triazine (168.3).

F. Synthesis of 5-(4-(4,6-dichloro-1,3,5 triazin-2ylamino)phenylamino)-2-amino-4-(4-fluorophenyl) pyridine-3-carbonitrile (5b):

A mixture of a compound (4b) (0.01 mole) dissolved in 40 ml ethanol and added malononitrile (0.01 mole), ammonium acetate

(0.08 mole) was refluxed for 8 hrs. Then the mixture was cooled and poured into crushed ice .The product (5b) separated out was filtered washed and recrystallized from alcohol. IR(KBR) cm^{-1} : C-Cl (776.208), Ar C-Cl (1129.12) ,Ar-N str (1380.78), primary N-H (1509.07), C=C (1626.66), N-H str (2918.73). ^1H NMR(CDCl_3) δ ppm : 4.296 to 4.314 (S,1H,s-triazine Ar-C-NH) , 4.331 (S,1H,Ar-C-NH₂), 6.986-7.437 (d, 4H,Ar-CH),7.588 (Ar-H), 9.898 (2-Py-Ar-1H). ^{13}C NMR(CDCl_3) δ ppm: Ar-CH (119.40 to 121.02), 2-Py-CH (134.15 to 135.28), S- triazine (168.20).

G. Synthesis of 5-(4-(4,6-dichloro-1,3,5 triazin-2ylamino)phenylamino)-2-amino-4-(benzo [d] [1,3] dioxol 4-yl pyridine-3-carbonitrile (5c):

A mixture of a compound (4c) (0.01 mole) dissolved in 40 ml ethanol and added malononitrile (0.01 mole), ammonium acetate (0.08 mole) was refluxed for 8 hrs. Then the mixture was cooled and poured into crushed ice .The product (5c) separated out was filtered washed and recrystallized from alcohol. C=C (1578.45), IR(KBR) cm^{-1} : C—Cl (813.61), C-O-C (1032.69), Ar C-Cl (1108.87) Ar-N (1334.50), primary N-H (1508.06), C=N (1616.06), N-H str (2922.59), O-H str (3784.62). ^1H NMR(CDCl_3) δ ppm : 4.427(S,1H,s-triazine Ar-C-NH) , 5.276(S,1H,Ar-C-NH₂), 6.672 (d,1H,Ar-Py), 6.983-7.469 (d, 4H,Ar-CH), ^{13}C NMR(CDCl_3) δ ppm: Ar-CH (108.82 to 121.53), 2-Py-CH (134.65 to 148.38), 1-imine (166.01), S- triazine (166.24).

III. RESULT AND DISCUSSION

The interest of organic chemistry in 2,4,6-trichloro-1,3,5-triazine as a starting material is due to temperature dependent reactivity of one chlorine atom that allow a sequential introduction of various substituents. In the present article we have reported the synthesis, characterization and antibacterial and Larvicidal activity of some novel s-triazine based cyanopyridine derivatives.

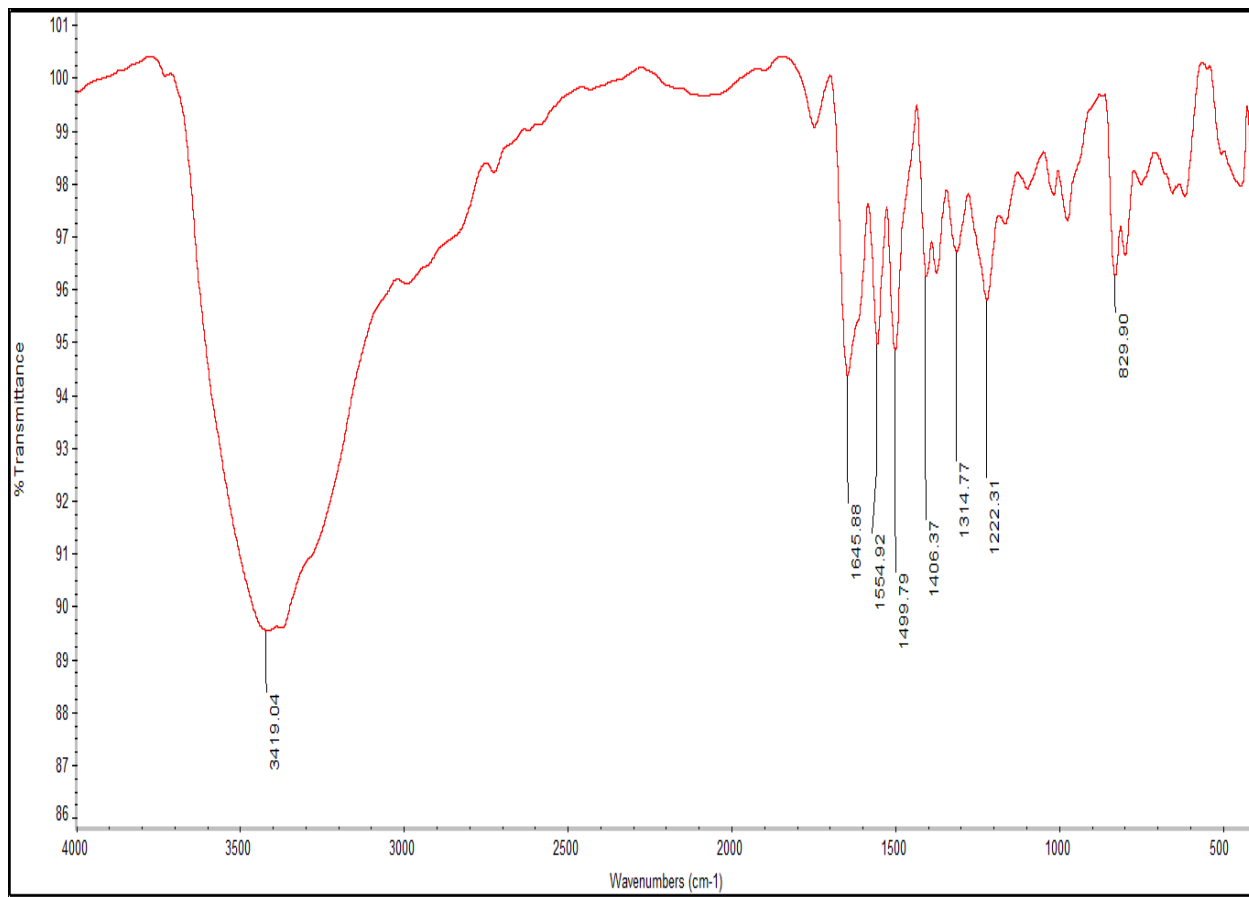
A. Antibacterial activity

All the synthesized compounds were screened for their antibacterial activity by using agar diffusion method against *S. aureus* and *E. faecalis* (Gram positive bacteria) and *E. coli*, *S. Typhi* (Gram negative bacteria) by using agar medium. Ciprofloxacin was used as standard drugs for the comparison of antibacterial activity by visualizing activity data it could be observed that compounds (5a-c) were found to be active or inactive against all bacterial strain. (Table No. 2)

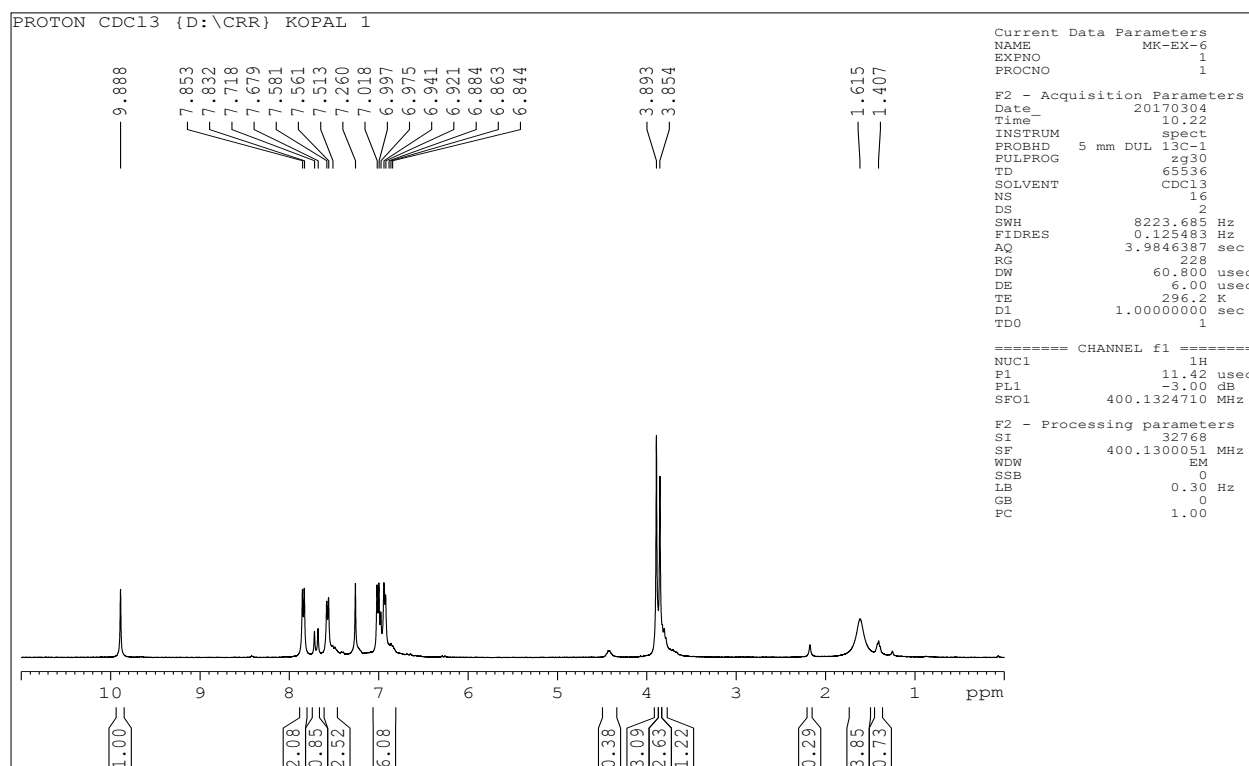
B. Larvicidal Activity

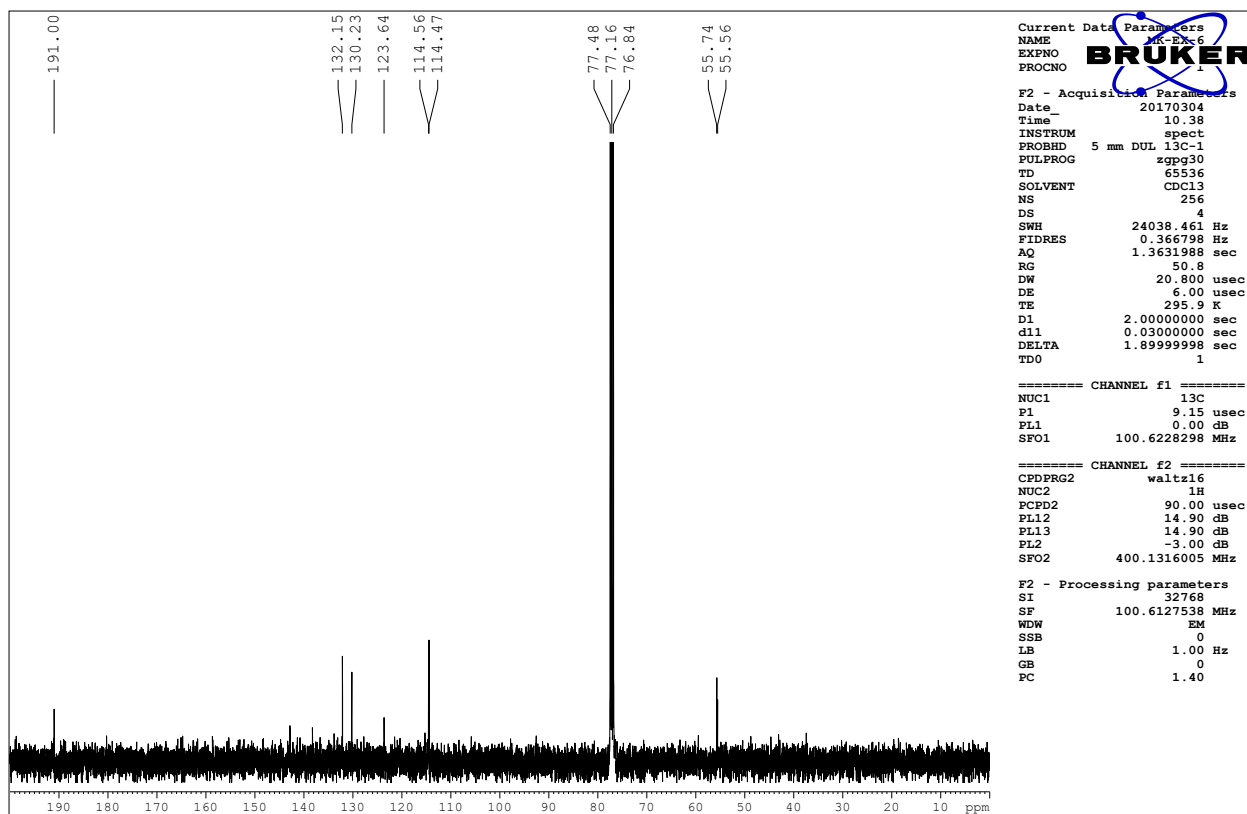
For the bioassay test, larvae were taken in five batches of 20 in 249 ml of water and 1.0 ml of the desired chemical extract concentration. The numbers of dead larvae were counted after 24 h of exposure and the percentage of mortality was reported from the average of five replicates.

C. Spectral Data

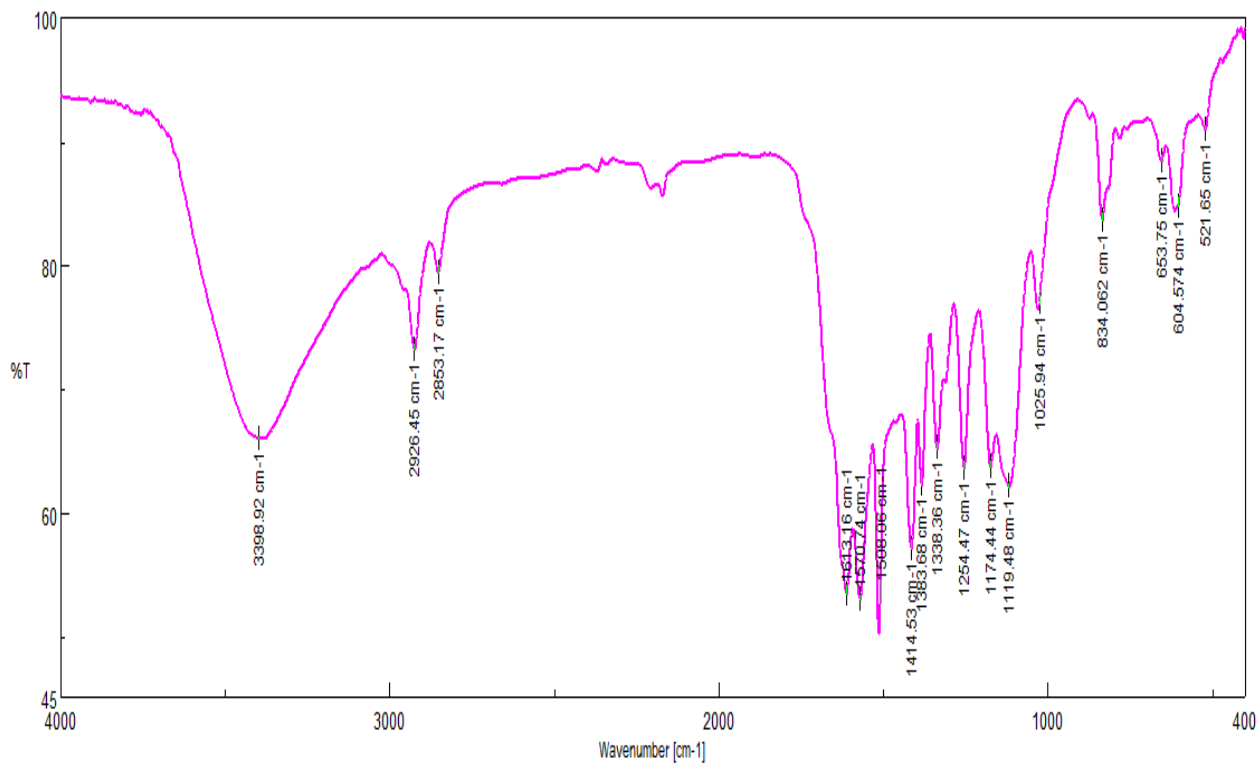


4a-¹H NMR

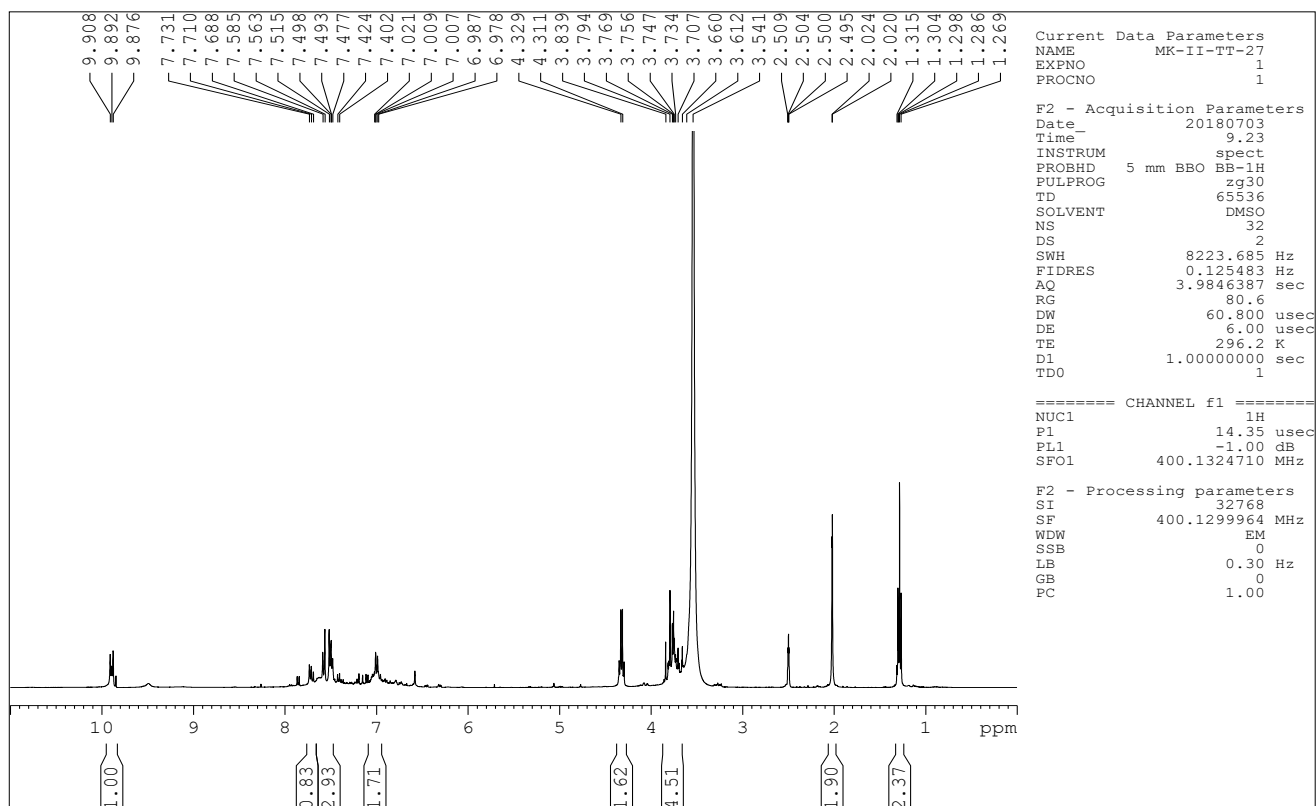




5a-I



5a-¹H-NMR



5A-¹³C-NMR

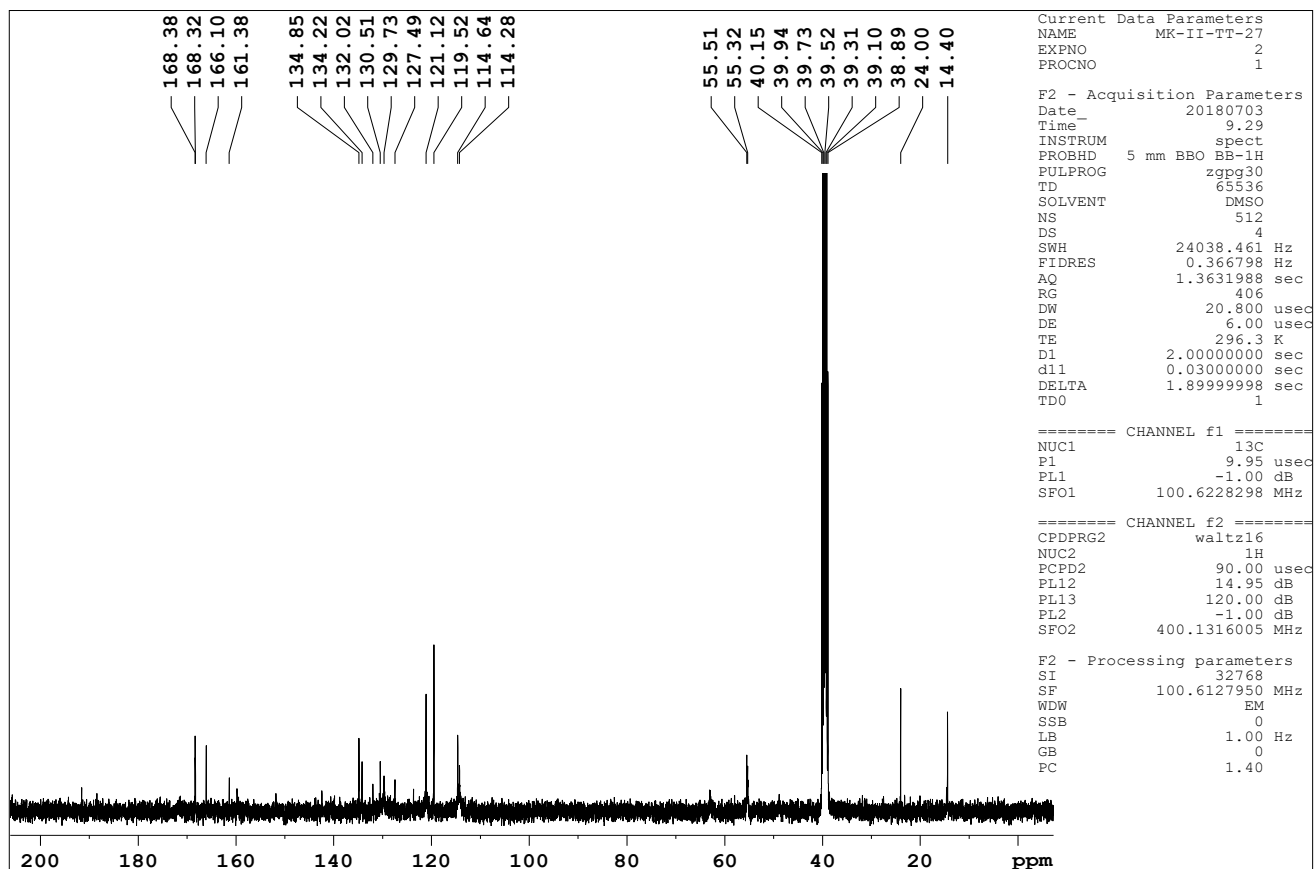


Table- 1: Physical data of the synthesized compounds (4a-c) and (5a-c)

Compound code	R	Mol. Formula	Mol. weight	MP(°C)	Yield(%)	Rf value
4a	C ₆ H ₄ OCH ₃	C ₁₉ H ₁₅ Cl ₂ N ₅ O ₂	416.26	190-191 °C	89%	0.61
4b	C ₆ H ₄ F	C ₁₇ H ₁₃ Cl ₂ FN ₅ O	404.23	194-196 °C	75%	0.70
4c	C ₇ H ₅ O ₂	C ₁₉ H ₁₃ Cl ₂ N ₅ O ₃	430.24	206-208° C	83%	0.53
5a	C ₆ H ₄ OCH ₃	C ₂₂ H ₁₆ Cl ₂ N ₈ O	479.32	115-120 °C	70 %	0.55
5b	C ₆ H ₄ F	C ₁₇ H ₁₃ Cl ₂ FN ₈	467.29	138-140° C	75%	0.65
5c	C ₇ H ₅ O ₂	C ₂₂ H ₁₄ Cl ₂ N ₈ O ₂	493.30	123-125° C	62 %	0.61

Table 2: Elemental analysis of the synthesized compounds (4a-c) and (5a-c)

Compound code	Mol. Formula	Appearance	Elemental Analysis		
			% C Calcd (found)	% H Calcd (found)	% N Calcd (found)
4a	C ₁₉ H ₁₅ Cl ₂ N ₅ O ₂	Light yellow	54.82 (54.80)	3.63 (3.60)	16.82 (16.21)
4b	C ₁₈ H ₁₂ Cl ₂ N ₅ OF	Half white	53.48 (53.46)	2.99 (2.97)	17.83 (17.80)
4c	C ₁₉ H ₁₃ Cl ₂ N ₅ O ₃	Pale yellow	53.04 (53.03)	3.05 (3.03)	16.28 (16.26)
5a	C ₂₂ H ₁₆ Cl ₂ N ₈ O	Greenish yellow	55.13 (55.10)	3.36 (3.33)	23.38 (23.35)
5b	C ₂₁ H ₁₃ Cl ₂ FN ₈	Dark brown	53.98 (53.95)	2.80 (2.78)	23.98 (23.97)
5c	C ₂₂ H ₁₄ Cl ₂ N ₈ O ₂	Brown	53.67 (53.66)	2.86 (2.84)	22.75 (22.73)

Table 3: Larvicidal activity

S.No	Chemical name with concentration	Effectiveness after 24 Hrs in % of killing
1	5a	79%
2	5b	80%
3	5c	72%

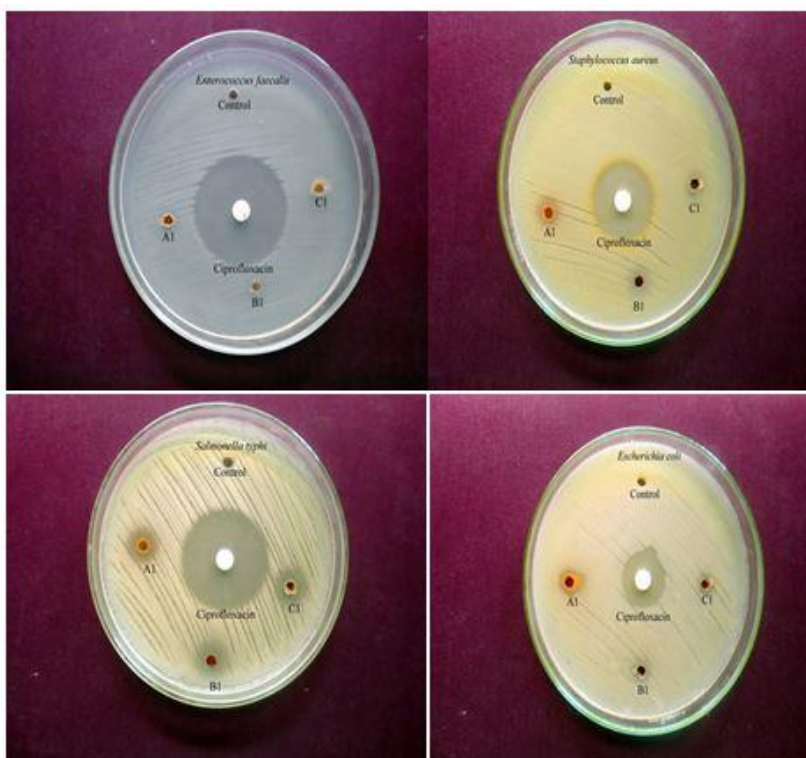
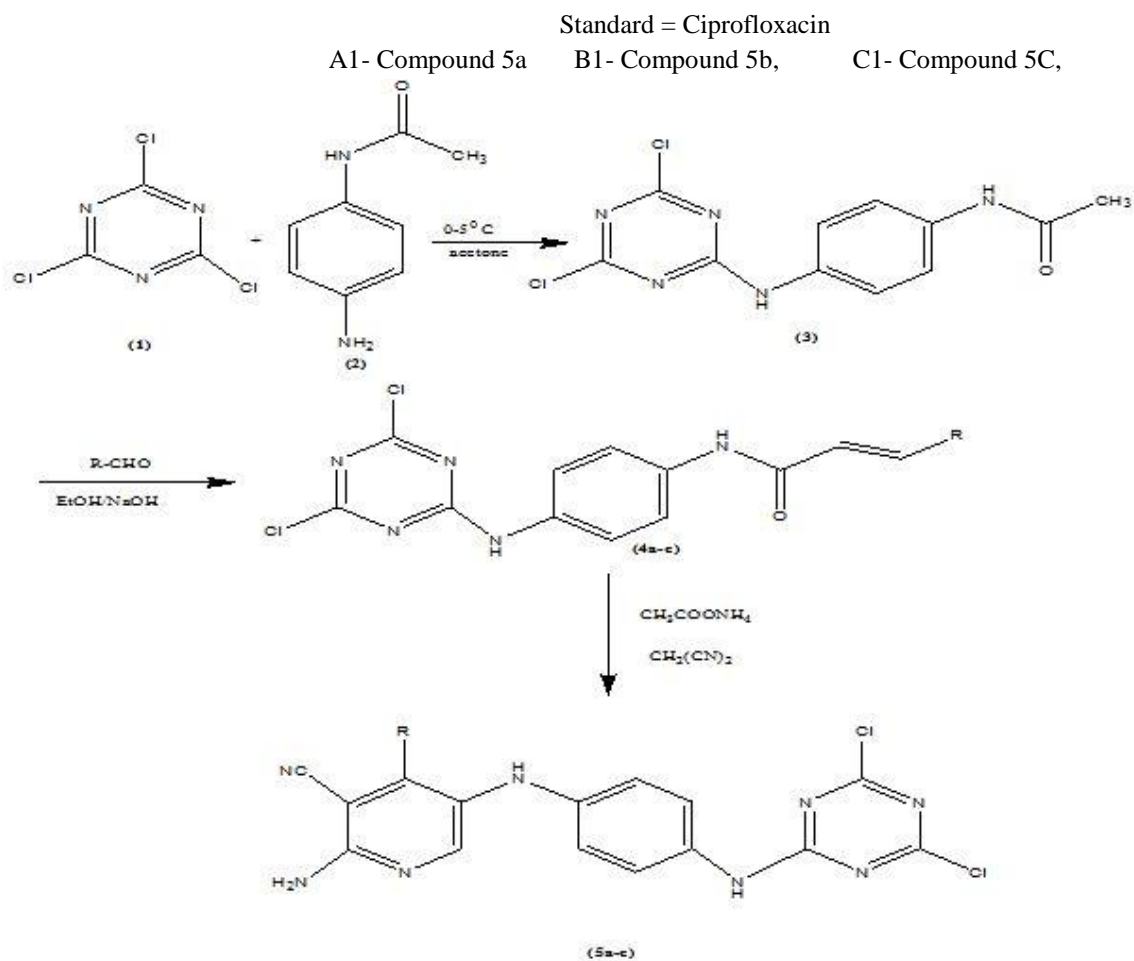


Figure 1. In vitro antibacterial activity data of s-triazine derivatives against tested organisms.



CONCLUSION

We have successfully synthesized a new series of chalcone derivatives and moreover, some compounds contain bioactive heterocyclic moiety. The antibacterial screening suggests that all the newly synthesized compounds showed moderate to good activity against the tested organism. The compounds showed good Larvicidal activity.

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